

## O 75: Overview Talk: Ulrike Diebold

Time: Thursday 9:30–10:15

Location: S054

**Invited Talk**

O 75.1 Thu 9:30 S054

**Ternary oxides with the perovskite structure exhibit an intriguingly rich variety in their physical and chemical properties.** — •ULRIKE DIEBOLD — Institute of Applied Physics, TU Wien, Wiedner Hauptstrasse 8-10/134, 1040 Vienna, Austria

The surfaces of these promising, yet complex materials are poorly understood. In the talk I will provide an overview of surface studies of perovskite oxides, and, for a few examples, discuss optimum preparation parameters for a reproducible surface structure and how this affects reactivity and growth.

The surface of  $\text{Sr}_3\text{Ru}_2\text{O}_7$ , the  $n=2$  member of the Ruddlesden-Popper series, is structurally quite simple: cleaving in UHV yields a  $\text{SrO}$ -like top layer, which is essentially defect-free, except for impurities

in the bulk material. The surface is very reactive towards components of the residual gas, however.  $\text{CO}$ ,  $\text{CO}_2$ , and  $\text{H}_2\text{O}$  adsorb readily and form adsorption complexes.  $\text{SrTiO}_3(110)$  is polar and exhibits a series of reconstructions that can be controlled by adjusting the chemical potential of its constituents, i.e. by evaporating appropriate amounts of Sr and Ti and annealing in  $\text{O}_2$ . The  $(n \times 1)$  reconstructions consist of a monolayer of titania with tetrahedrally-coordinated Ti atoms that are arranged in corner-sharing rings. When the Ti chemical potential is increased, the surface switches over to a  $(2 \times m)$  symmetry with a titania layer that is composed of Ti in octahedral coordination. These stoichiometry-dependent, facile structural changes have a profound effect on surface reactivity, and on the homoepitaxial growth of  $\text{SrTiO}_3$  during pulsed laser deposition.