## Monday

## CPP 27: Focus Session: When theory meets experiment: Hybrid halide perovskites for applications beyond solar II (joint session HL/CPP)

Hybrid halide perovskites are by now well established solar absorber and emitter materials, with power conversion efficiencies of single cell devices exceeding 20

Organizers: Linn Leppert (Universiät Bayreuth) and Felix Deschler (TU Munich)

Time: Monday 15:00-17:00

Invited Talk CPP 27.1 Mon 15:00 POT 251 Double perovskite electronic structures: A chemical perspective — •ADAM SLAVNEY<sup>1</sup>, HEMAMALA KARUNADASA<sup>2</sup>, LINN LEPPERT<sup>3</sup>, and BRIDGET CONNOR<sup>2</sup> — <sup>1</sup>Department of Chemistry, Harvard University, Cambridge, Massachusetts, USA — <sup>2</sup>Department of Chemistry, Stanford University, Stanford, California, USA — <sup>3</sup>Institute of Physics, University of Bayreuth, Bayreuth, Germany

Halide double perovskites, of the formula A2BB'X6, are close analogs to the APbX3 lead halide perovskites and have attracted significant interest as possible non-toxic alternatives to the lead materials in perovskite photovoltaics. Double perovskites divide the formally +2charge on the octahedral B site unevenly over two B sites, allowing for cations with charges from +1 - +4 to be incorporated into the perovskite lattice. I will discuss the results of our experimental and theoretical investigations into double perovskites over the last several years, with a particular emphasis on understanding the differences in electronic structure between single and double perovskites. The double perovskite electronic structure is largely dictated by the combination of the B and B\* cations rather than either cation individually and I will provide simple rules which accurately predict the band structure from the chemical formula in nearly every case. Double perovskites electronic structures have features not available in single perovskites including symmetry-forbidden bandgaps and unusually small bandgaps generated by metal-metal charge transfer transitions. Finally, I will discuss some of the defect chemistry of double perovskites and how those lessons can be applied to halide perovskites more broadly.

Invited Talk CPP 27.2 Mon 15:30 POT 251 Solid state ionics of hybrid halide perovskites: equilibrium situation and light effects — •ALESSANDRO SENOCRATE<sup>1,2</sup>, GEE YEONG KIM<sup>1</sup>, TAE YOUL YANG<sup>1</sup>, GIULIANO GREGORI<sup>1</sup>, MICHAEL GRAETZEL<sup>1,2</sup>, and JOACHIM MAIER<sup>1</sup> — <sup>1</sup>Max Planck Institute for Solid State Research, Stuttgart, Germany — <sup>2</sup>École polytechnique fédérale de Lausanne

In recent years, hybrid halide perovskites have been attracting great attention due to their exceptional photo-electrochemical properties. When used as light-harvesters in solar cells, device efficiencies exceeding 25% can be realized. We showed that a deeper understanding of (i) functionality, (ii) stability, as well as (iii) the possibility to improve the performance require a thorough insight into non-stoichiometry and ion transport.

In this contribution, we study the nature of the ionic conductivity in methylammonium lead iodide (MAPbI<sub>3</sub>), the archetypal halide perovskite, by means of a great number of electrochemical and nuclear magnetic techniques. To aid the experimental investigation, we include detailed defect chemical modelling describing the effects of varying iodine partial pressure (stoichiometry) and dopant content. By extending this study to the situation under illumination, we observe a striking enhancement of ionic conductivity by more than 2 orders of magnitude in MAPbI<sub>3</sub>, alongside the expected increase in electronic conductivity. We provide a mechanistic explanation of this astonishing phenomenon and discuss its relevance for the photo-stability of this class of materials.

## CPP 27.3 Mon 16:00 POT 251

Looking beyond the surface: The band gap of bulk methylammonium lead iodide — OSKAR SCHUSTER<sup>1</sup>, PETER WIENTJES<sup>1</sup>, SHREETU SHRESTHA<sup>2</sup>, IEVGEN LEVCHUK<sup>2</sup>, MYKHAILO SYTNYK<sup>3</sup>, GEBHARD MATT<sup>2</sup>, ANDRES OSVET<sup>2</sup>, MIROSLAW BATENTSCHUK<sup>2</sup>, WOLFGANG HEISS<sup>3</sup>, CHRISTOPH BRABEC<sup>2</sup>, THOMAS FAUSTER<sup>1</sup>, and •DANIEL NIESNER<sup>1</sup> — <sup>1</sup>Lehrstuhl für Festkörperphysik, Friedrich-Alexander-Universität Erlangen-Nürnberg (FAU), Staudtstr. 7, 91058 Erlangen, Germany — <sup>2</sup>Materials Science Department (I-MEET), FAU, Martensstrasse 7, 91058 Erlangen, Germany — <sup>3</sup>Materials Science Department, FAU, Energy Campus Nürnberg, 90429 Nürnberg, Germany Reported optical properties of lead halide perovskites as basic as the absorption onset and the band gap vary significantly. To unambiguously answer the question whether the discrepancies are a result of differences between bulk and "near-surface" material, we perform bulk-sensitive two-photon absorption with an information depth in the millimeter range on single crystals. In the application-relevant room-temperature tetragonal phase (170 K) we find a bulk band gap of  $1.55 \pm 0.01$  eV. Reference measurements on the surface-near region in a reflection geometry show excitonic transitions at  $1.59 \pm 0.01$  eV, consistent with the literature. The small band gap of the bulk material explains the extended infrared absorption of crystalline perovskite so-lar cells, the narrow bandwidth of crystalline perovskite photodetectors making use of the spectral filtering at the surface, and the low-energy bands which carry optically driven spin-polarized currents.

CPP 27.4 Mon 16:15 POT 251 Double Peak Emission in Lead Halide Perovskites by Self-Absorption — •KONSTANTIN SCHÖTZ<sup>1</sup>, ABDELRAHMAN M. ASKAR<sup>2</sup>, WEI PENG<sup>3</sup>, DOMINIK SEEBERGER<sup>1</sup>, TANAJI P. GUJAR<sup>1</sup>, MUKUN-DAN THELAKKAT<sup>1</sup>, SVEN HUETTNER<sup>1</sup>, OSMAN M. BAKR<sup>3</sup>, KARTHIK SHANKAR<sup>2</sup>, ANNA KÖHLER<sup>1</sup>, and FABIAN PANZER<sup>1</sup> — <sup>1</sup>Universität Bayreuth, Bayreuth, Germany — <sup>2</sup>University of Alberta, Edmonton, Canada — <sup>3</sup>King Abdullah University of Science and Technology (KAUST), Thuwal, Kingdom of Saudi Arabia

Despite the rapidly increasing efficiencies of perovskite solar cells, the optoelectronic properties of this material class are not completely understood. Especially when measured photoluminescence (PL) spectra consist of multiple peaks, their origin is still debated. In this work, we investigate in detail double peak PL spectra of halide perovskite thin films and single crystals with different material compositions. By different optical spectroscopic approaches and quantitative models, we demonstrate that the additional PL peak results from an extensive self-absorption effect, whose impact is intensified by strong internal reflections. This self-absorption accounts for the unusual temperature dependence of the additional PL peak and it implies that absorption until far into the Urbach tail of the perovskite is important. The strong internal reflections entail that even for thin films self-absorption can have a significant contribution to the PL spectrum. Our results allow for a clear assignment of the PL peaks by differentiating between optical effects and actual electronic transitions, which is necessary for understanding the optoelectronic properties of halide perovskites.

CPP 27.5 Mon 16:30 POT 251 Surface atomic and electronic structure of CsPbBr<sub>3</sub> inorganic perovskite — •JANEK RIEGER, FELIX TRUNK, THOMAS FAUSTER, and DANIEL NIESNER — Lehrstuhl für Festkörperphysik, Friedrich-Alexander University Erlangen-Nürnberg (FAU), Staudtstr. 7, D-91058 Erlangen, Germany

The crystallographic orientation of the surface has a strong impact on the performance of solar cells built from lead-halide perovskites [1]. For a systematic study of the surface carrier-dynamics, we prepare atomically well-defined (001) surfaces of the inorganic perovskite CsPbBr<sub>3</sub> by epitaxial growth in ultra-high vacuum. Low-energy electron diffraction was carried out to analyze the resulting samples.

The measured surface unit cell as well as the corresponding phase transitions differ from the bulk-terminated ones. Angle-resolved photoelectron spectroscopy finds a valence-band structure in line with data we obtained from single crystals. Finally, we also investigated the energetics of the conduction-band electrons using two-photon photoelectron spectroscopy. From the positions of the valence-band maximum and the conduction-band minimum we conclude that our CsPbBr<sub>3</sub> samples are intrinsic.

[1] S. Y. Leblebici et al., Nat. Energy 1.8, 16093 (2016).

CPP 27.6 Mon 16:45 POT 251 Finite temperature effects on the excited states in layered

Location: POT 251

 ${\bf BiI}_3 - {\bullet}{\bf I}{\rm VONA}$  BRAVIC and BARTOMEU MONSERRAT - TCM Group, Cavendish Laboratory, University of Cambridge, J. J. Thomson Avenue, Cambridge CB3 0HE, United Kingdom

Over the last decade layered  $\operatorname{BiI}_3$  has been matter of investigation in the optoelectronics community as it exhibits a fundamental band gap similar to that of silicon but behaves like a direct band gap semiconductor. However, the excited state properties of this system remained poorly understood, which is mostly a consequence of the extremely flat band structure and the arising low robustness with respect to dynamical distortions, volume expansion and stacking faults. In order to determine its potential performance in an optoelectronic device at finite temperature, it proves invaluable to establish a fundamental understanding of its excited states and how the aforementioned properties effect those. In this study we reveal the effect of spin-orbit coupling, electron-phonon coupling, exciton-phonon coupling, stacking faults and volume expansion on the optical response and photoluminescence properties of this materials using DFT as well as GW-BSE, proving that it is crucial to account for all these effects to provide a comprehensive understanding of the optoelectronic performance of layered BiI<sub>3</sub>, and in general layered van-der Waals materials.