## MO 11: Photochemistry I

MO 11.1 Wed 10:30 MO-H7 UV and Mid-IR Photo-induced Dissociation of Solvated (Bio)Molecular Complexes - •Mukhtar Singh ${ }^{1,2,3}$, Matthew Scott Robinson ${ }^{1,2,3}$, Hubertus Bromberger ${ }^{1,2}$ Jolijn Onvlee $^{1,3}$, Sebastian Trippel ${ }^{1,2}$, and Jochen $\mathrm{Küpper}^{1,2,3}$ - ${ }^{1}$ Center for Free-Electron Laser Science, Deutsches ElektronenSynchrotron DESY, Hamburg - ${ }^{2}$ Center for Ultrafast Imaging, Universität Hamburg - ${ }^{3}$ Department of Physics, Universität Hamburg
We present the ultrafast imaging of UV and thermal energy chemical dynamics of micro-solvated (bio)molecular complexes probed with strong field techniques [1]. We produced a pure sample of the molecule of interest in the gas phase by using a combination of a molecular beam and the electrostatic deflector [2]. To study the induced dynamics, we set up the both, an UV and a mid-IR pump-probe experiment, in which a 266 nm and $2.9 \mu \mathrm{~m}$ beam was used to excite the system. A $1.3 \mu \mathrm{~m}$ beam was used for ionising the system. First experiments focused on the ion imaging of the UV and mid-IR triggered system. Future experiments will use laser-induced electron diffraction (LIED) [3,4] to probe the induced dynamics in order to obtain structural information about the system with atomic resolution.
[1] J. Onvlee, et al., arXiv:2103.07171v1 12 Mar 2021.
[2] S. Trippel, et al., Rev. Sci. Instrum. 89, 096110 (2018).
[3] J. Wiese, et al., Phys. Rev. Research 3, 013089 (2021).
[4] E. T. Karamatskos, et al., J. Chem. Phys. 150, 244301(2019).
MO 11.2 Wed 10:45 MO-H7
Multiphoton light-induced potentials - $\bullet$ Matthias Kübel Institut für Optik und Quantenelektronik, FSU Jena
We study the strong-field photodissociation of $\mathrm{H}_{2}^{+}$using Cold Target Recoil Ion Momentum Spectroscopy. Our two-color streaking method allows us to produce a coherent wave packet in the molecular cation and expose it to a phase-controlled mid-infrared laser field. The resulting fragmentation pattern exhibits a strikingly structured angular distribution, which depends on various laser parameters. Using two-color Floquet theory as well as numerical solutions of the time-dependent Schrödinger equation, we show that the pattern arises from an interplay of competing multiphoton fragmentation pathways, as well as forced rotational dynamics. On a qualitative level, the dynamics can be intuitively understood by picturing the light-induced potential energy landscape on which the nuclear motion takes place. Interestingly, our expeirmental approach allows us to shape these potentials and follow the ensuing molecular dynamics. Our results highlight the complexity of intense matter interaction even in the simplest of molecules.

MO 11.3 Wed 11:00 MO-H7 A Velocity Map Imaging Study of the Photodissociation Dynamics of the Trichloromethyl Radical - •Christian Matthaei $^{1}$, Deb Pratim Mukhopadhyay ${ }^{1,2}$, and Ingo Fischer ${ }^{1}$ - ${ }^{1}$ Institute of Physical and Theoretical Chemistry, University of Würzburg, Am Hubland, D-97074 Würzburg - ${ }^{2}$ Present address: Department of Dynamics of Molecules and Clusters, J. Heyrovský Institute of Physical Chemistry, Dolejškova 2155/3, 18223 Praha 8, Czech Republic
$\mathrm{CCl}_{3}$ is one of the numerous halogen-containing molecules that contribute to the catalytic destruction of the ozone layer. Here we investigate the photodissociation dynamics of this molecule following the
excitation with light between 250 and 230 nm . We mainly observe the loss of a Cl atom, that is associated with the $\mathrm{CCl}_{2}$ fragment. However, the generation of CCl could also be observed. Control experiments using $\mathrm{CCl}_{2}$ as a precursor suggest that the CCl results from the reaction $\mathrm{CCl}_{3}$ to $\mathrm{CCl}+\mathrm{Cl}_{2}$.

MO 11.4 Wed 11:15 MO-H7
Mechanistic Studies on a Deracemization Reaction via a Triplet 1,3-Diradical Induced by Energy Transfer from a Chiral Sensitizer - •Roger Jan Kutta ${ }^{1}$, Xinyao $\mathrm{Li}^{2}$, Christian Jandl ${ }^{2}$, Andreas Bauer ${ }^{2}$, Thorsten Bach ${ }^{2}$, and Patrick Nuernberger ${ }^{1}$ - ${ }^{1}$ Institut für Physikalische und Theoretische Chemie, Universität Regensburg, Germany - ${ }^{2}$ Lehrstuhl für Organische Chemie I, Technische Universität München, Germany
The photochemical deracemization[1] of spiro[cyclopropane-1,3'-indolin]-2'-ones (spirocyclopropyl oxindoles) was investigated by timeresolved spectroscopies and computational approaches. The corresponding 2,2-dichloro compound is configurationally labile upon direct irradiation and when sensitized by excited achiral thioxanthen-9-one. In the latter reaction the triplet 1,3-diradical intermediate is generated via triplet energy transfer from the photosensitizer.

Deracemization is achieved by using a chiral thioxanthone photosensitizer with a lactam hydrogen bonding site. Here, three factors co-act favorably for high enantioselectivity: i) a factor 3 differing binding constants to the chiral thioxanthone for the two enantiomers. ii) unequal molecular distances in the complexes, presumably lead to differing energy transfer efficiencies. iii) the 1,3-diradical lifetime exceeds the complex lifetime, facilitating a racemic deactivation back to the ground state.[2]
[1] Nature 2018, 564, 240-243.
[2] Angew. Chem. Int. Ed. 2020, 59, 21640-21647.
MO 11.5 Wed 11:30 MO-H7
Selenyl Radicals in Solution: Photogeneration and Reactions - •Carina Allacher ${ }^{1}$, Roger Jan Kutta ${ }^{1}$, Elias Harrer $^{1}$, Daniel Grenda ${ }^{1}$, Amit Kumar-Dutta ${ }^{2}$, Sooyoung Park $^{2}$, Alexander Breder ${ }^{2}$, and Patrick Nuernberger ${ }^{1}$ ${ }^{1}$ Institut für Physikalische und Theoretische Chemie, Universität Regensburg, 93040 Regensburg - ${ }^{2}$ Institut für Organische Chemie, Universität Regensburg, 93040 Regensburg
Due to the low bond energy of selenium bonds, selenium-centered radicals can be formed photochemically with visible or near-UV light [1], thereby opening up new synthetic strategies based on the versatile reactivity of these radical species. A prominent precursor of the phenylselenyl radical $\left(\mathrm{PhSe}^{\bullet}\right)$ is diphenyl diselenide $\left(\mathrm{Ph}_{2} \mathrm{Se}_{2}\right)$ which upon irradiation with UV light may dissociate homolytically [2]. In this work, we identify an additional reaction pathway that is pursued after excitation of $\mathrm{Ph}_{2} \mathrm{Se}_{2}$ with 355 nm light. To quantify the dynamics of these competing photochemical processes, transient absorption studies on a timescale from nano- to milliseconds are performed in various solvent environments. Solely in the solvent hexafluoro-2-propanol, a further intermediate, namely the radical $\mathrm{PhSeH}^{+\bullet}$, can be photogenerated. Beyond $\mathrm{Ph}_{2} \mathrm{Se}_{2}$, we present time-resolved studies on further organoselenium compounds as light-triggered sources of $\mathrm{PhSe}{ }^{\bullet}$ and $\mathrm{PhSeH}^{+\bullet}$ radicals.
[1] S. Ji et al., Angew. Chem. Int. Ed. 53, 6781 (2014).
[2] I. P. Beletskaya et al., J. Chem. Soc., Perkin Trans. 2, 107 (2007).

